

ENT-KAURANES AND TRACHYLOBANES FROM *HELIANTHUS RADULA*

WERNER HERZ and PALIANAPPAN KULANTHAIVEL

Department of Chemistry, The Florida State University, Tallahassee, FL 32306, U.S.A.

(Received 11 February 1983)

Key Word Index—*Helianthus radula*; Compositae; Heliantheae; ent-kauranes; trachylobanes; 11-oxotrachyloban-19-oic acid; ester of (–)-16 α -hydroxykauranoic acid and 11 α -hydroxytrachyloban-19-oic acid.

Abstract—The aerial parts of *Helianthus radula* afforded a variety of known ent-kauranoic acids as well as two new substances, 11-oxotrachyloban-19-oic acid and an ester involving the carboxyl group of (–)-16 α -hydroxykauranoic acid and the hydroxyl group of 11 α -hydroxytrachyloban-19-oic acid.

INTRODUCTION

In continuation of our earlier work on *Helianthus* species [1–6] we have examined *Helianthus radula* (Pursh) T. & G. which is a species found in the pine barrens of the deep South [7]. This contained various ent-kauranoic acids as do many other previously studied *Helianthus* species [3, 5, 6, 8–13], but gave none of the sesquiterpene lactones which have been found in various other representatives of the genus [1–4, 6, 8, 9, 11, 13–22]. Two new trachylobanes were also isolated.

RESULTS AND DISCUSSION

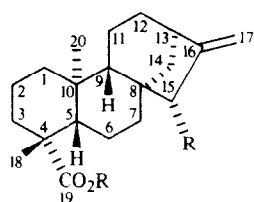
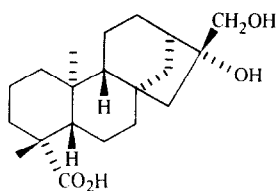
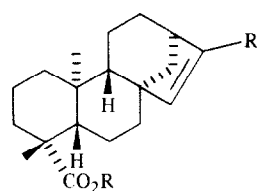
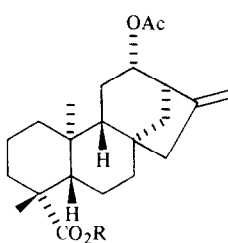
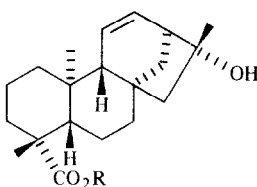
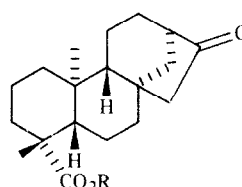
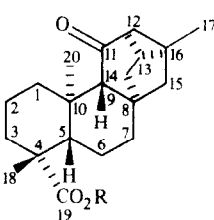
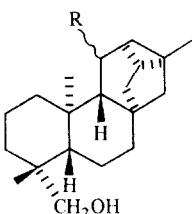
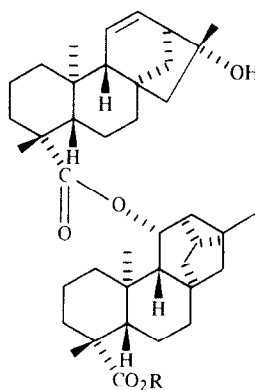
The main diterpenoid constituent of the chloroform extract of *H. radula* was **1a**; other ent-kauranes isolated in smaller amounts were **1b**, **2**, **3a**, **3c**, **4a** [8], the last three in the form of methyl esters **3b**, **3d** and **4b**, and **5a**. The nor-ent-kauranone **6a**, isolated in the form of its methyl ester **6b**, was identified through its ¹H NMR (see Experimental) and ¹³C NMR spectra (Table 2). Compound **6a** has been obtained previously by degradation of **1a** [23], but never to our knowledge as a natural product.

11-Oxotrachyloban-19-oic acid (**7a**) was isolated in the form of its methyl ester **7b**, which could not be induced to crystallize. The empirical formula C₂₁H₃₀O₃ (high resolution MS), IR spectrum (C=O frequencies at 1725 and 1675 cm^{–1}) and ¹H NMR and ¹³C NMR spectra (Tables 1 and 2) characterized it as the methyl ester of a ketonic trachyloban-19-oic acid.* In the ¹H NMR spectrum the unusual chemical shift of a signal at δ 2.83, which was assigned to H-1 β by decoupling experiments, could be explained only by placing the carbonyl group at C-11 (model). In the ¹³C NMR spectrum the significant downfield shifts of the signals associated with C-9 and the cyclopropyl carbons C-12, C-13 and C-16, compared with the spectra of trachyloban-19-oic acid [29], ciliaric acid [13] and related compounds [30, 31], were also in accord

with this formulation, as was the IR band at 1675 cm^{–1} which indicated conjugation and inclusion of the carbonyl in a 6-membered ring or larger. The location of the keto group at C-11 was further proved by LiAlH₄ reduction (NaBH₄ was ineffective) of **7b** to a 3:2 mixture of diols **8a** and **8b**. The multiplicities and coupling constants of the protons under the new secondary hydroxyl group (δ 4.53 dd, $J = 9, 4$ Hz for **8a**, δ 4.11 tbr, $J_s = 3, 3$ Hz for **8b**) were as expected for 11 β - and 11 α -hydroxylated derivatives (models). The CD curve of **7b**, which exhibited a positive Cotton effect ($[\theta]_{298} + 4360$), indicated that the absolute configuration was as depicted in the formula.

A very minor constituent was the acid **9a**, again isolated as a methyl ester, mp 233–234°, which exhibited IR bands at 3450, 1730 and 1700 cm^{–1}. The molecular formula C₄₁H₆₀O₅ was established by high resolution mass spectrometry; detailed analysis of the ¹H NMR (see Experimental) and ¹³C NMR spectra (Table 2) led to formula **9b**. The presence of the ent-16 β -hydroxykaur-11-en-19-oate fragment (A half) could be inferred from the IR and ¹H NMR spectra, with the latter displaying the characteristic low field C-17 methyl and mutually coupled vinylic proton signals at δ 5.91 and 5.55 of this moiety [5], and was verified by comparing the ¹³C NMR spectra of **5b** and **9b** (Table 2). The ¹H NMR spectrum of **9b** also exhibited a low field signal at δ 5.28, apparently due to hydrogen under an ester function in the second (B) half of the molecule, as a second double bond was lacking and as the δ 65–90 region of the ¹³C NMR spectrum exhibited only one signal, a doublet at δ 71.11, in addition to the C-16 singlet of the A half. The appearance (tbr) and coupling constants ($J = 3, 3$ Hz) of this signal were strictly comparable to H-11 β of **8b**; this and the appropriate shifts in the ¹H NMR and ¹³C NMR spectra identified this portion of the molecule as methyl 11 α -hydroxytrachyloban-19-oate. Paucity of material prevented verification of the structure by chemical methods. However, the alternative formula, in which the ether linkage between the two halves involves the carboxyl group of the trachylobane and the tertiary hydroxyl of the kaurane portion, is clearly ruled out by the chemical shifts in rings C and D of both halves and that of H-11 in the B half. In a recent revision [32] of Heiser's infrageneric

* Trachyloban-19-oic acids have been encountered in several other *Helianthus* species [3, 5, 10, 11, 13, 24–27] as well as in members of the related genus *Viguiera* [28].

**1a** R=H**1b** R=OAng**2****3a** R=H, R'=CHO**3b** R=Me, R'=CHO**3c** R=H, R'=CH₂OH**3d** R=Me, R'=CH₂OH**4a** R=H**4b** R=Me**5a** R=H**5b** R=Me**6a** R=H**6b** R=Me**7a** R=H**7b** R=Me**8a** R=β-OH**8b** R=α-OH**9a** R=H**9b** R=Me

classification of *Helianthus* [7], *H. radula* was placed in series *Angustifolii* of section *Divaricati*. Chemically it more closely resembles *H. rigidus* [5] and *H. occidentalis* [6] of series *Atrorubentes* in producing *ent*-kaurenic acids and trachylobanic acids but no lactones. It shows less resemblance to *H. angustifolius* [8, 9] and *H. simulans* [6], the only other previously studied members of series *Angustifolii*, which yielded *ent*-kaurenic acids, very similar lactones and the flavone hymenoxin.

EXPERIMENTAL

Above-ground parts of *Helianthus radula* (Pursh) T. & G. (1.1 kg) collected by Mr. D. Gage in the vicinity of Tallahassee in September 1979 (voucher not available) was extracted with CHCl₃ and worked up in the usual manner. The crude gum on agitation with CHCl₃ gave a residue (0.8 g) identified as the usual mixture of sitosterol and stigmasterol β-D-glucosides. The mother liquor was evapd, the crude gum (28 g) was adsorbed on

Table 1. ^1H NMR spectral data of compound **7b***

Proton No.	7b	7b + Eu (fod) ₃
H-1a	2.83 <i>t</i> db (3, 13)	3.04
1b	1.10 <i>m</i>	1.18
3a	2.15 <i>t</i> db (3, 13)	2.18
3b	1.03 <i>dt</i> (4, 13)	1.06
5	1.10 <i>m</i>	1.13 <i>dd</i> (12, 3)
6a	1.87 <i>qd</i> (3, 13)	obsc.
9	1.41 <i>s</i>	1.78 <i>s</i>
14a	2.27 <i>db</i> (12)	2.38
14b	1.49 <i>db</i>	obsc.
15a	1.67 <i>d</i> (12)	obsc.
15b	1.37 <i>d</i> (12)	obsc.
17†	1.26	1.28
18†	1.17	1.19
20†	0.78	0.88
OMe†	3.63	3.65

*Run at 270 MHz in CDCl_3 with TMS as internal standard. Chemical shifts in δ . Coupling constants in parentheses in hertz. Signals of unlisted protons were obscured (obsc.).

†Intensity of three protons.

40 g silica gel (Mallinckrodt 100 mesh) and chromatographed over 0.3 kg of the same adsorbent in hexane, 500 ml fractions being collected as follows: 1–4 (hexane), 5–10 (hexane–EtOAc, 19:1), 11–16 (hexane–EtOAc, 9:1), 17–22 (hexane–EtOAc, 4:1), 23–26 (hexane–EtOAc, 3:2), 27–30 (hexane–EtOAc, 2:3), 31–34 (hexane–EtOAc, 1:4) 35–38 (EtOAc), 39–40 (EtOAc–MeOH,

99:1), 41–42 (EtOAc–MeOH, 49:1), 43–44 (EtOAc–MeOH, 19:1) and 45–46 (EtOAc–MeOH, 9:1).

Fraction 6 (0.605 g) on purification by TLC (hexane–EtOAc, 19:1) furnished 32 mg of **1a**. Fraction 7 (6.3 g) was mainly **1a**. Purification of fractions 9, 11 and 12 by TLC (C_6H_6 –EtOAc, 39:1) gave 0.22 g of a mixture of α -amyrin, β -amyrin and lupeol, 66 mg of **1b**, mp 198–200°, and 72 mg of a mixture of sitosterol and stigmasterol. Fraction 17 (0.645 g) on initial purification gave a mixture of **3a** and **4a**. Methylation with CH_2N_2 and subsequent purification by TLC (C_6H_6 –EtOAc, 39:1) yielded 19 mg of **3b**, 53 mg of **4b** and 43 mg of a mixture of **3b** and **4b**. Methylation of fractions 18 and 19 (combined wt 0.54 g) and purification by TLC (C_6H_6 –EtOAc, 39:1, developed $\times 3$) gave **3b** and **4b** in 5 mg amounts and **7b** in 67 mg yield.

Trituration of fractions 23 and 24 with hexane–EtOAc gave 441 mg of **5a**, mp 241–247° (EtOAc–MeOH). Trituration of fraction 30 with Et_2O gave 42 mg of **2**, mp 268–270°, fractions 21 and 22 and the mother liquors of fractions 23 and 24 were combined (wt 2.6 g), esterified with CH_2N_2 and purified by a combination of CC (silica gel) and TLC (C_6H_6 –EtOAc, 39:1, multiple developments) to yield 37 mg of **6b**, 9 mg of **9b**, 84 mg of **3d**, mp 130–132°, and 170 mg of **5b**. Compound **6b** had mp 143–145°, lit. mp. 142–143° [23], IR bands (KBr) 1742 and 1725 cm^{-1} ; ^1H NMR (270 MHz, CDCl_3): δ 2.40 (*m*, H-13), 2.20 (*db*, $J = 13$ Hz, H-3a), 1.21 (H-18) and 0.91 (H-20). The ^{13}C NMR spectrum is listed in Table 2.

Methyl 11-oxotrachyloban-19-oate (**7b**). Compound **7b** which could not be induced to crystallize had IR bands (neat) at 1725 and 1675 cm^{-1} ; CD curve (MeOH) $[\theta]_{298} + 4360$ (maximum); ^1H NMR and ^{13}C NMR spectra are listed in Tables 1 and 2. (Calc. for $\text{C}_{21}\text{H}_{30}\text{O}_3$: MW, 330.2195. Found: MW (MS), 330.2167.) Other significant ions in the high resolution MS were at m/z (rel. int.): 315 ($\text{C}_{20}\text{H}_{27}\text{O}_3$, 2.1), 271 ($\text{C}_{19}\text{H}_{27}\text{O}$, 9.5), 270

Table 2. ^{13}C NMR spectral data for compounds **5b**, **6a**, **7b** and **9b***

Carbon No.	5b	6a	7b	9b	
				A half	B half
C-1	41.30 <i>t</i> †	41.09 <i>t</i> †	41.36 <i>t</i>	41.48 <i>t</i>	39.88 <i>t</i>
2	19.00 <i>t</i>	18.76 <i>t</i>	18.77 <i>t</i>	18.89 <i>t</i> , 18.55 <i>t</i>	
3	38.01 <i>t</i>	37.97 <i>t</i>	37.85 <i>t</i> †	38.29 <i>t</i> , 37.92 <i>t</i>	
4	43.24 §	43.82	43.94	43.65†	43.65
5	55.83 <i>d</i>	56.84 <i>d</i>	56.82 <i>d</i>	55.64 <i>d</i>	56.53 <i>d</i>
6	21.86 <i>t</i>	20.82 <i>t</i>	21.32 <i>t</i>	22.19 <i>t</i> , 21.98 <i>t</i>	
7	39.87 <i>t</i> †	40.71 <i>t</i> †	38.31 <i>t</i>	39.50 <i>t</i>	38.48 <i>t</i>
8	43.85 §	42.46	39.82 §	43.25†	40.12
9	50.06 <i>d</i>	54.03 <i>d</i>	65.55 <i>d</i> †	50.12 <i>d</i>	59.31 <i>d</i>
10	38.68	39.56	39.74	38.89, 38.11	
11	130.50 <i>d</i>	19.08 <i>t</i>	211.56	132.86 <i>d</i> †	71.11 <i>d</i>
12	126.93 <i>d</i>	29.49 <i>t</i>	40.23 <i>d</i>	126.91 <i>d</i> †	24.59 <i>d</i>
13	61.16 <i>d</i>	47.77 <i>d</i>	31.12 <i>d</i>	61.20 <i>d</i>	23.74 <i>d</i>
14	34.39 <i>t</i>	37.32 <i>t</i>	34.20 <i>t</i>	34.51 <i>t</i>	32.38 <i>t</i>
15	58.59 <i>t</i>	54.98 <i>t</i>	48.55 <i>t</i>	58.60 <i>t</i>	48.71 <i>t</i>
16	83.75	222.37	31.24	83.71	21.98
17	25.81 <i>q</i>	—	19.38 <i>q</i>	25.77 <i>q</i>	20.15 <i>q</i>
18	28.53 <i>q</i>	28.76 <i>q</i>	29.87 <i>q</i>	28.73 <i>q</i> , 28.63 <i>q</i>	
19	177.83	177.83	177.57	176.38	177.42
20	15.31 <i>q</i>	15.92 <i>q</i>	14.90 <i>q</i>	15.94 <i>q</i>	13.68 <i>q</i>
OMe	51.18 <i>q</i>	51.21 <i>q</i>	51.18 <i>q</i>	—	51.11 <i>q</i>

*Run in CDCl_3 at 67.98 MHz with TMS as internal standard. Unmarked signals are singlets.

†Assignments made by selective irradiation.

‡, §Assignments may be interchanged.

(C₁₉H₂₆O, 5.4) and 255 (C₁₈H₂₃O, 5.8).

Reduction with NaBH₄ in MeOH resulted in the recovery of **7b**. Reduction of 20 mg of **7b** in 1 ml THF with 20 mg LiAlH₄ for 30 min at 45° followed by the usual work-up gave 12 mg of a 3:2 mixture of diols **8a** and **8b**. (Calc. for C₂₀H₃₂O₂: MW, 304.2402. Found: MW (MS), 304.2388.) Other significant ions were at *m/z* (rel. int.): 286 (C₂₀H₃₀O, 32.4), 273 (C₁₉H₂₉O, 44.5), 255 (C₁₉H₂₇, 100). The major diol **8a** exhibited ¹H NMR signals (270 MHz, CDCl₃) at δ4.53 (*dd*, *J* = 9, 4 Hz, H-11α), 3.69 (*d*, *J* = 11 Hz, H-19α) and 3.45 (*dd*, *J* = 11 Hz, H-19b). The isomeric diol **8b** had signals at δ4.11 (*tbr*, *J* = 3 Hz, H-11β), 3.69 (*d*, *J* = 11 Hz, H-19a) and 3.43 (*dd*, *J* = 11 Hz, H-19b).

Ester 9b. Compound **9b**, mp 233–234° (EtOAc), IR ν_{\max}^{KBr} cm⁻¹: 3450, 1730 and 1700; ¹H NMR (270 MHz, CDCl₃): δ5.91 (*ddbr*, *J* = 10, 8 Hz) and 5.55 (*dd*, *J* = 10, 4 Hz, H-12 and H-11 of kaurene half), 5.28 (*tbr*, *J* = 3 Hz, H-11β of trachylobane half) 2.20 (*m*, H-13 of kaurene half), 1.56 (H-17 of kaurene half), 1.16 (H-18 of kaurene, H-17 and H-18 of trachylobane), 0.85 (H-20 of trachylobane) and 0.76 (H-20 of kaurene). The ¹³C NMR spectrum is listed in Table 2. (Calc. for C₄₁H₆₀O₅: MW, 632.4437. Found: MW (MS), 632.4447.) The low resolution MS also exhibited significant ions at *m/z* (rel. int.): 6.14 (0.2), 331 (0.4), 317 (2.6), 315 (100), 299 (11.9), 273 (5.4) and 255 (91.9).

Acknowledgement—This work was supported in part by a grant (CA-13121) from the U.S. Public Health Service through the National Cancer Institute.

REFERENCES

- Herz, W. and De Groote, R. (1977) *Phytochemistry* **16**, 1307.
- Herz, W. and Kumar, N. (1981) *Phytochemistry* **20**, 93.
- Herz, W. and Kumar, N. (1981) *Phytochemistry* **20**, 99.
- Herz, W. and Kumar, N. (1981) *Phytochemistry* **20**, 1339.
- Herz, W., Govindan, S. V. and Watanabe, K. (1982) *Phytochemistry* **21**, 946.
- Herz, W., Kulanthaivel, P. and Watanabe, K. (1983) *Phytochemistry* **22**, 2021.
- Heiser, C. B. (1966) *Mem. Torrey Bot. Club* **22**, No. 3.
- Bohlmann, F., Jakupovic, J., King, R. M. and Robinson, H. (1980) *Phytochemistry* **19**, 863.
- Ohno, N., Gershenzon, J., Neuman, P. and Mabry, T. J. (1981) *Phytochemistry* **20**, 2393.
- Ferguson, G., McCrindle, R., Murphy, S. T. and Parvez, M. (1982) *J. Chem. Res. (Suppl.)* 200.
- Watanabe, K., Ohno, N., Yoshioka, H., Gershenzon, J. and Mabry, T. J. (1982) *Phytochemistry* **21**, 709.
- Ohno, N., Mabry, T. J., Zabel, V. and Watson (1979) *Phytochemistry* **18**, 1687.
- Ohno, N. and Mabry, T. J. (1980) *Phytochemistry*, **19**, 609.
- Spring, O., Albert, K. and Gradmann, W. (1981) *Phytochemistry* **20**, 1883.
- Spring, O., Albert, K. and Hager, A. (1982) *Phytochemistry* **21**, 2551.
- Ortega, A., Romo de Vivar, A., Diaz, E. and Romo, J. (1970) *Rev. Latinoam. Quim.* **1**, 81.
- Bohlmann, F. and Dutta, L. N. (1979) *Phytochemistry* **18**, 676.
- Ohno, N. and Mabry, T. J. (1979) *Phytochemistry* **18**, 1003.
- Iriuchima, S., Kuyama, S., Takahashi, N. and Tamura, S. (1966) *Agric. Biol. Chem. (Tokyo)* **30**, 511.
- Morimoto, H., Sonno, Y. and Oshio, H. (1966) *Tetrahedron* **22**, 3173.
- Nishikawa, M., Kamiya, K., Takabatake, A. and Oshio, H. (1966) *Tetrahedron* **22**, 3601.
- Morimoto, H. and Oshio, H. (1981) *J. Nat. Prod. (Lloydia)* **44**, 748.
- Jefferies, P. R. and Payne, T. G. (1965) *Aust. J. Chem.* **18**, 1441.
- Pyrek, J. St. (1970) *Tetrahedron* **26**, 5029.
- Kasprzyk, Z., Janiszewka, W. and Papaj, M. (1974) *Bull. Acad. Pol. Sci. Ser. Sci. Biol.* **22**, 1.
- Bjeldanes, L. F. and Geissman, T. A. (1970) *Phytochemistry* **11**, 327.
- Ortega, A., Ayala, A., Guerrero, C. and Romo de Vivar, A. (1972) *Rev. Soc. Quim. Mex.* **16**, 191.
- Bohlmann, F., Jakupovic, J., Ahmed, M., Grenz, M., Suding, H., Robinson, H. and King, R. M. (1981) *Phytochemistry* **20**, 113.
- Cory, R. M. and Stothers, J. B. (1978) *Org. Magn. Reson.* **11**, 252.
- Arnone, A., Mondelli, R. and Pyrek, J. St. (1979) *Org. Magn. Reson.* **12**, 429.
- Hasan, C. M., Healey, T. M. and Waterman, P. G. (1982) *Phytochemistry* **21**, 177.
- Schilling, E. E. and Heiser, C. B. (1981) *Taxon* **30**, 393.
- Herz, W. and Högenauer, G. (1962) *J. Org. Chem.* **27**, 905.